Pressure and Atmospheric Gases

■ Partial Pressure of gases: *PV=nRT*

• According to Dalton's Law: $P_{\text{TOT}}V = V(P_1 + P_2 + ... + P_n)$, so partial pressure of gas $x(P_x)$ is: $P_x = (P_{\text{TOT}} \times \% x)/100$

• In dry air at sea level, $P_{\text{TOT}} = 760 \text{ mm Hg}$, and:

 $N_2 = 78.1\%$ $P_{\rm N_2} = 594 \,\rm mm \, Hg$

 $O_2 = 20.95\%$ $P_{O_2} = 159 \text{ mm Hg}$

- CO₂ = 0.03% $P_{\rm CO_2} = 0.23 \,\rm mm \, Hg$
- Have to factor in water vapor pressure $(P_{H,O})$:

 $P_{\text{TOT}}V = V (P_1 + P_2 + \dots + P_n + P_{H_2O})$

Example: in saturated air at 20 °C, 760 mm Hg, P_{H_2O} = 17.5 mm

Hg. P_{O_2} therefore = 0.2095(760 - 17.5) = 155 mm Hg

Standard Temperature and Pressure (STP)

- STP is 0 °C (273 K) and 760 mm Hg
- Frequently used to present data in 'common' form Example: dry air at (20.95% O₂) @ 20 °C and 740 mm Hg has 210 ml O₂/l @ STP. At STP:
- $(273 \text{ K}/293 \text{ K}) \times (740 \text{ mm Hg}/760 \text{ mm Hg}) = 190 \text{ ml } O_2/l$ ■ Therefore, mass/vol ∝ moles/vol ∝ vol @ STP/vol. But,
- when dealing with volumes, must specify temperature and pressure
- At constant pressure, $[gas x] \propto P_x$, increasing temperature decreases gas x
- The absorption coefficient (A) = volume in solution @ STP/l water. Influenced by temperature, salinity
- At equilibrium, P_x in water = P_x in air above it (gas tension)

Partial Pressures and Blood

■ Partial pressures determine equilibria: relative concentrations in air and water don't matter

P_{O_2} in air	$P_{\rm CO_2}$ in air	
↓↑	↓↑	
P_{O_2} in solution	$P_{\rm CO_2}$ in solution	
in blood	in blood	
↓↑	↓↑	
not HbO ₂	not H ₂ CO ₃	
	$\int \uparrow \\ \text{not HCO}_3^- \int Don't part in equilibrium of HCO_3^-$	icipate rium

Oxygen Availability in Different Media

	For ml (STP) l^{-1} at $P_{O_2} = 159$ mm Hg:		
Medium	0 °C	12 °C	24 °C
Air	208	200	192 (↓ 8%)
Distilled H ₂ O	10.2	7.7	6.2 (40%)
Seawater	8.0	6.1	4.9 (↓ 52%)
		(0.1	

■ Air is a (MUCH!) richer source of O₂ than water.

- Because organisms have to respire a certain amount of O2, these differences determine how much air or water must pass over their respiratory surfaces Need 20x more water than air by volume
- Differences become apparent greater as temperature \uparrow . At P_{O_2} = 159 mm Hg and 0 °C, to obtain 1 l of O2 requires passing 4.8 l of air, 98 l of distilled water, or 125 l of seawater over respiratory surface.
- Air breathing therefore supports the highest MRs, aquatic environments limited by low $[O_2]$. Ectotherms especially limited, as MR[†] with [†]T

Gaseous Diffusion in Different Media

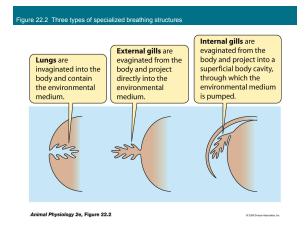
■ Rate of O₂ diffusion in air is 300,000× greater than in water Organisms compensate by:

- Moving through water ('ram ventilation')
- Pumping water over gills
- Staying in moving water
- Rate of CO₂ diffusion in air is less than O₂, rate of CO₂ diffusion in water is about 25x greater than O₂ Air TA7-4---

	AII	water
Density (g/ml @ 17 °C)	0.0012	about 1 (800× greater)

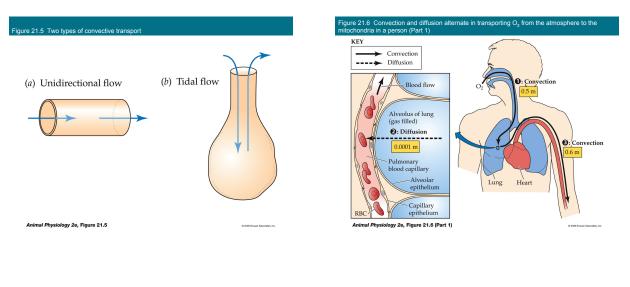
Viscosity	0 °C	Water about 100× greater than air			
	40 °C	Water about 35× greater than air			
Increased density and viscosity means that more energy must					
be expended to move water over respiratory surfaces					
<i>Example:</i> humans use 1-2% of RMR for ventilation					

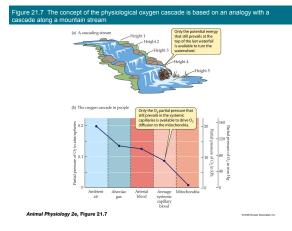
fishes use 10-20% of RMR for ventilation

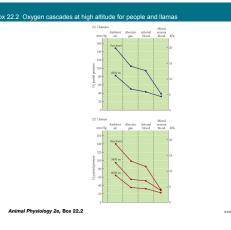


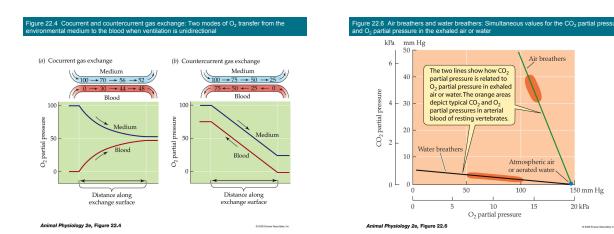
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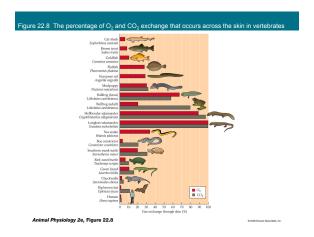




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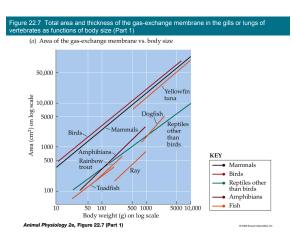
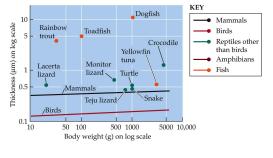
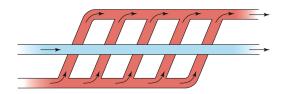


Figure 22.7 Total area and thickness of the gas-exchange membrane in the gills or lungs of vertebrates as functions of body size (Part 2)

(b) Thickness of the gas-exchange membrane vs. body size



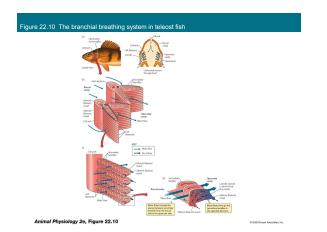
Animal Physiology 2e, Figure 22.7 (Part 2)

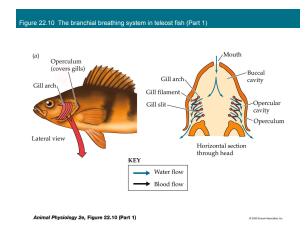


e 22.5 Cross-current gas exchange: A third mode of O2 transfer from the environmenta um to the blood when ventilation is unidirectional

Animal Physiology 2e, Figure 22.5

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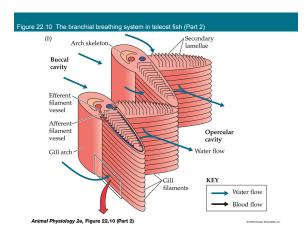




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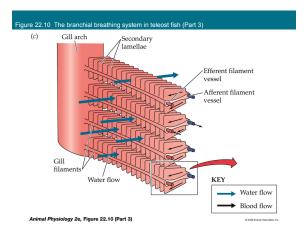
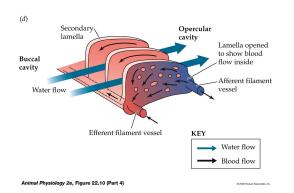


Figure 22.10 The branchial breathing system in teleost fish (Part 4)

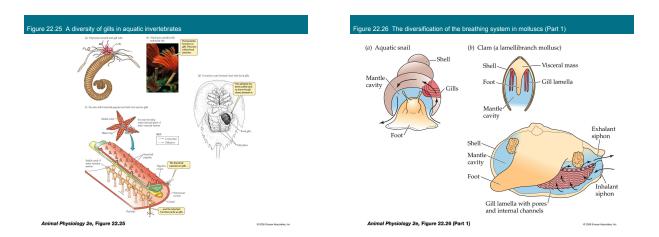


Chapter 22 Opener Throughout the animal kingdom, species that depend on vigorous endurance



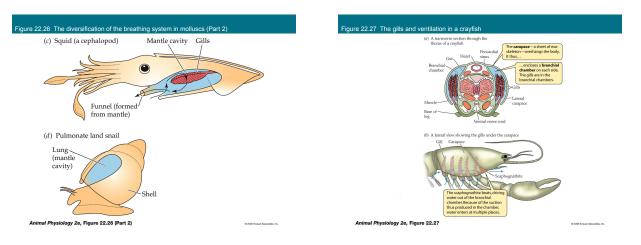
Animal Physiology 2e, Chapter 22 Opener

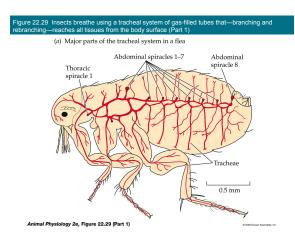
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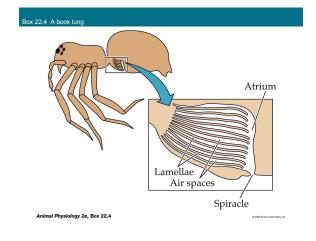


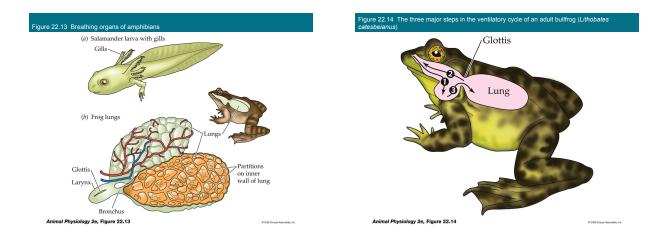
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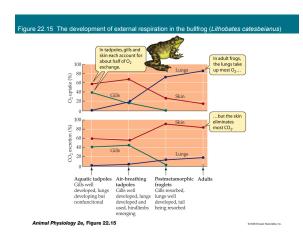


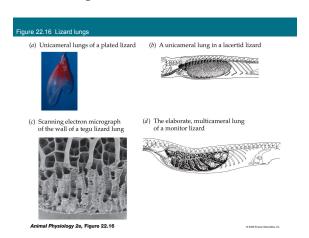


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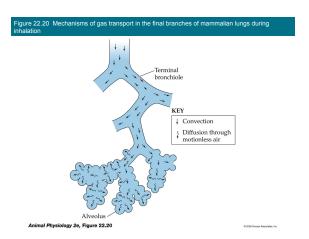
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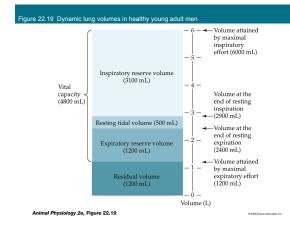
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gure 22.18 Respiratory aiways of the mammalian lung (Part 1) (a) The finest airways of the mammalian lung, ending in alveoli Alveolar duct Terminal Respiratory Norchioles Alveolus Alveolar sac





Avian Respiratory Systems

- Differs radically from the lungs of both reptiles and mammals.
- Most efficient respiratory system of any vertebrate • The finest branches of the bronchi do not
- terminate in alveoli, but are tube-like *parabronchi*.
- Lung Posterior
- A large portion of the inspired air bypasses the lungs and flows directly to the air sacs. • On expiration, this oxygenated air flows through the lungs allowing continuous
- air flow. • It takes two respiratory cycles for a single breath of air to pass through the
- system.
- Air sacs extend into the thorax, abdomen, and
- even into the long bones.
- This air sac system also helps cool birds during vigorous flying, when up to 27 times more heat is produced.
- The air sacs in the bones, legs and wings can also provide considerable buoyancy to the bird (the usually warmer air acts like a hot air balloon).

Comparative Animal Physiology Gas Exchange and Breathing

